

SYNTHESIS OF 12a-DEOXY-5a,6-ANHYDROTETRACYCLINE.

THE FIRST TOTAL SYNTHESIS

OF THE NATURALLY OCCURRING TETRACYCLINE (1)

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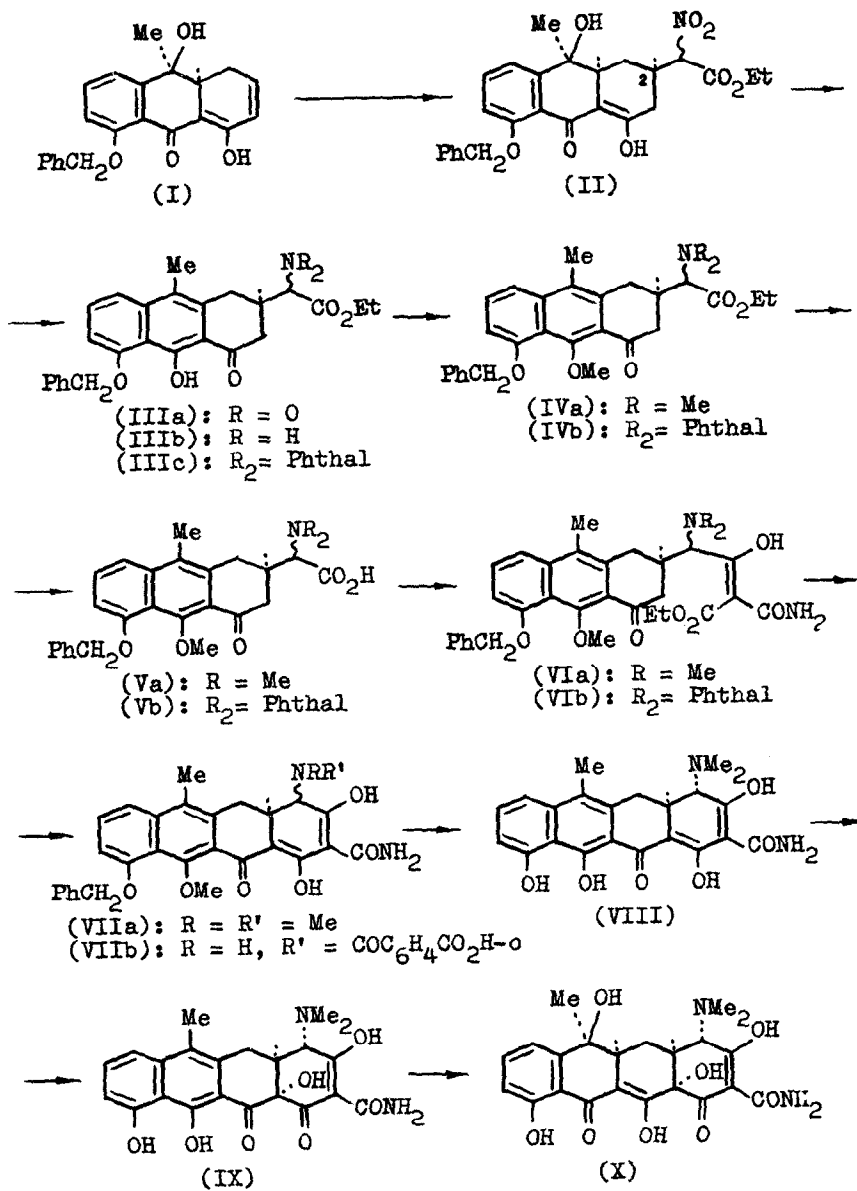
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IN the process of studies on hydronaphthacene compounds, closely related to anhydrotetracyclines, we have synthesized ( $\pm$ )-12a-deoxy-5a,6-anhydrotetracycline (VIII). Since we had 12a-hydroxylated the (-)-component of racemate (VIII) into 5a,6-anhydrotetracycline (IX) (2) that had earlier been 5a,6-hydrated into tetracycline (X) by Schach von Wittenau (3) this accomplishes the first total synthesis of a naturally occurring tetracycline antibiotic, and, namely, tetracycline (X), itself.

The starting point in the synthesis was the dienediolone (I) prepared in 6 stages from juglone (4). Its condensation with the triethylammonium salt of ethyl nitroacetate in tetrahydrofuran gave a mixture of C<sub>2</sub>-epimeric adducts (II) (IIa: m.p. 134-135°, from EtOH;  $\lambda_{\max}$  (here and further in EtOH) 278, 340 m $\mu$  ( $\lg \epsilon$  3.77, 4.17);  $\nu_{\max}$  (here and further in Nujol)



3400, 1752, 1615, 1593, 1561, 1376  $\text{cm}^{-1}$ . IIb: m.p. 140-141°, from EtOH;  $\lambda_{\text{max}}$  277, 340  $\text{m}\mu$  ( $\lg \epsilon$  3.76, 4.16);  $\nu_{\text{max}}$  3490, 1742, 1610, 1593, 1560, 1376  $\text{cm}^{-1}$ ). On treatment with 0.4 N alcoholic HCl at 80° the adducts are dehydrated into the nitro-compound (IIIa) (m.p. 126-128°, from EtOH;  $\lambda_{\text{max}}$  221, 269, 297, 309, 321, 408  $\text{m}\mu$  ( $\lg \epsilon$  4.50, 4.65, 3.69, 3.65, 3.50, 3.98);  $\nu_{\text{max}}$  2680, 1750, 1616, 1577, 1380  $\text{cm}^{-1}$ ), which is smoothly reduced by zinc dust in acetic acid to the amino ester (IIIb) (m.p. 138-140°, from EtOH;  $\lambda_{\text{max}}$  223, 269, 297, 309, 322, 408  $\text{m}\mu$  ( $\lg \epsilon$  4.62, 4.81, 3.85, 3.80, 3.64, 4.15);  $\nu_{\text{max}}$  3400, 3340, 2690, 1740, 1615, 1577  $\text{cm}^{-1}$ ).

In order to synthesize deoxyanhydrotetracycline (VIII) via the dimethylamino derivatives (VIa) and (VIIa) we first methylated the amino ester (IIIb) by MeI + Ag<sub>2</sub>O in dimethylformamide to the dimethylamino ester (IVa) (m.p. 106-108°, from EtOH;  $\lambda_{\text{max}}$  222, 262, 307, 318, 373  $\text{m}\mu$  ( $\lg \epsilon$  4.65, 4.72, 3.84, 3.74, 3.96);  $\nu_{\text{max}}$  1720, 1675, 1609, 1580  $\text{cm}^{-1}$ ) and saponified the latter to dimethylamino acid (Va) ( $\lambda_{\text{max}}$  223, 268, 307, 318, 366  $\text{m}\mu$  ( $\lg \epsilon$  4.49, 4.43, 3.72, 3.65, 3.72);  $\nu_{\text{max}}$  3450, 2700, 1684, 1615, 1582  $\text{cm}^{-1}$ ). However attempts to condense the acid (Va) as the chloride, the isopropyl carbonate (cf. (5)) or the isobutyl carbonate with ethyl ethoxymagnesium malonamate did not yield the desired acylmalonamate (VIa).

We, therefore, acylated the amino ester (IIIb) with carboethoxyphthalimide in tetrahydrofuran to obtain the phthaloyl derivative (IIIc) (m.p. 176-178°, from EtOH;  $\lambda_{\text{max}}$  222, 269, 297, 308, 322, 407  $\text{m}\mu$  ( $\lg \epsilon$  4.60, 4.78, 3.52, 3.50, 3.18, 3.87);  $\nu_{\text{max}}$  1780, 1750, 1720, 1618, 1580  $\text{cm}^{-1}$ ) which was then methylated by MeI + Ag<sub>2</sub>O to (IVb) (m.p. 166-167°, from benzene;  $\lambda_{\text{max}}$  221, 263, 307, 320, 375  $\text{m}\mu$  ( $\lg \epsilon$  4.48, 4.74, 3.93, 3.76, 3.96);  $\nu_{\text{max}}$  1785, 1753, 1735, 1715, 1692, 1612, 1580  $\text{cm}^{-1}$ ). Saponification of this compound with 0.1 N KOH in aqueous tetrahydrofuran followed by recyclization of the phthalimido

grouping by heating in diglyme at 140° gave the acid (Vb)(m.p. 202-204°, from MeNO<sub>2</sub>;  $\lambda_{\max}$  222, 262, 307, 318, 374  $\mu$  (lg  $\epsilon$  4.63, 4.72, 3.81, 3.71, 3.92);  $\nu_{\max}$  3480, 3360, 1778, 1727, 1610, 1580  $\text{cm}^{-1}$ ), which was then converted into the substituted ethyl N-phthaloylglycylmalonamate (VIb) ( $\lambda_{\max}$  223, 261, 308, 319, 374  $\mu$  (lg  $\epsilon$  4.42, 4.68, 3.75, 3.56, 3.89);  $\nu_{\max}$  3350, 1780, 1727, 1687, 1665, 1610, 1585  $\text{cm}^{-1}$ ) by treatment with PCl<sub>5</sub> + HCONMe<sub>2</sub> (cf. (6)) in tetrahydrofuran, followed by EtOMgCH(CO<sub>2</sub>Et)CONH<sub>2</sub>. The compound (VIb) was cyclized by means of MeSOCH<sub>2</sub>Na in dimethylsulfoxide into the substituted hydro-naphthacene (VIIb)(m.p. 170-180°;  $\lambda_{\max}$  224, 266, 297, 310, 320, 388  $\mu$  (lg  $\epsilon$  4.67, 4.57, 3.94, 3.91, 3.84, 4.02);  $\nu_{\max}$  3350, 2700, 1712, 1695, 1680, 1662, 1615, 1603, 1583  $\text{cm}^{-1}$ ) and this on hydrolysis with HBr in AcOH and methylation with MeI in tetrahydrofuran gave ( $\pm$ )-12a-deoxy-5a,6-anhydrotetracycline (VIII) identified spectroscopically and chromatographically with the (-)-isomer prepared by degradation of the naturally occurring tetracycline according to the procedure of Green and Boothe (7).

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